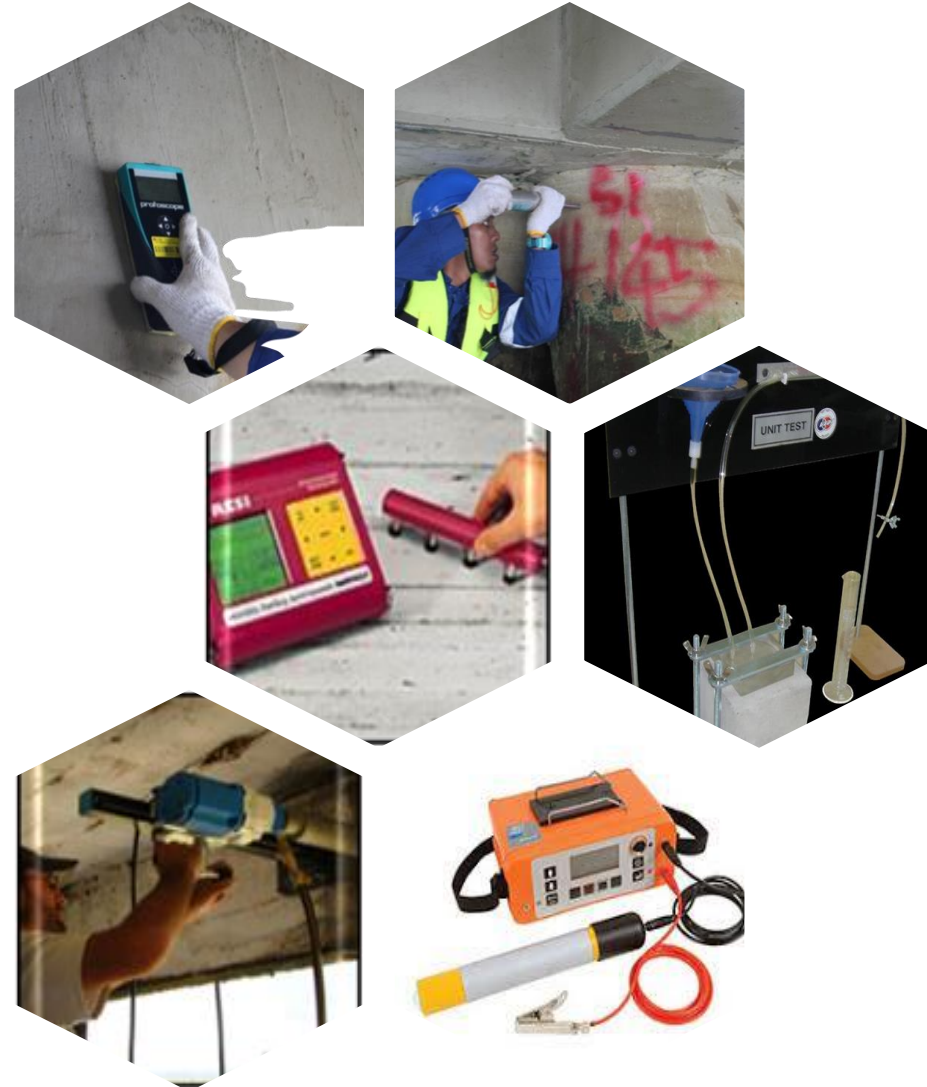


# ASSESSMENT METHODS



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**Research interest:** sustainable waste material, concrete technology, Lightweight concrete, green concrete, **Concrete durability**, sustainable development, sustainable energy production, Geopolymer concrete, Building Information Modelling (BIM).





What test that you used to check corrosion of steel reinforcement and deterioration of concrete?

Go to [www.menti.com](https://www.menti.com) and use the code 6154 5071

# Assessment

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## Strength

- Assessment on concrete strength

## Integrity

- Determination of localised integrity or generalised assessment of behaviour of whole structure.

## Durability

- Identifying nature & extent of observed or suspected deterioration including reinforcement corrosion.

# Assessment



## Strength

- Coring
- Windsor Probe
- Pull-Out
- Pull-Off



## Integrity

- Rebound Hammer
- Ultrasonic Pulse Velocity
- Thermography
- Radiography



## Durability

- Carbonation
- Chloride content
- Sulphate content

# Corrosion

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- Measurement of steel reinforcement corrosion in concrete is **essential** to analyze the strength and durability of a structure.
- Steel corrosion **reduce the life span** of the structure but also **increases the cost** for inspection and maintenance significantly.
- Currently, **no instrument or technique** is available which can measure the extent of corrosion of steel.
- However, measurement of concrete properties, such as **resistivity** and **half-cell potential** of concrete can assess the **probability** of corrosion of reinforcing steel.

# Basis for Corrosion Measurement

- ions penetrate into concrete through **pores** which render concrete to act as an **electrolyte**.
- As electrons move in concrete, an **electrical potential field** inside concrete over steel bars is generated.
- The majority of corrosion measurement techniques are based on examining the **electrochemical condition** of the rebar-concrete interface.
- This examination is generally conducted from the **surface of concrete**.
- Different techniques have been developed and used for measurement of steel corrosion, for instance- **Resistivity meter and Half-cell potential**.

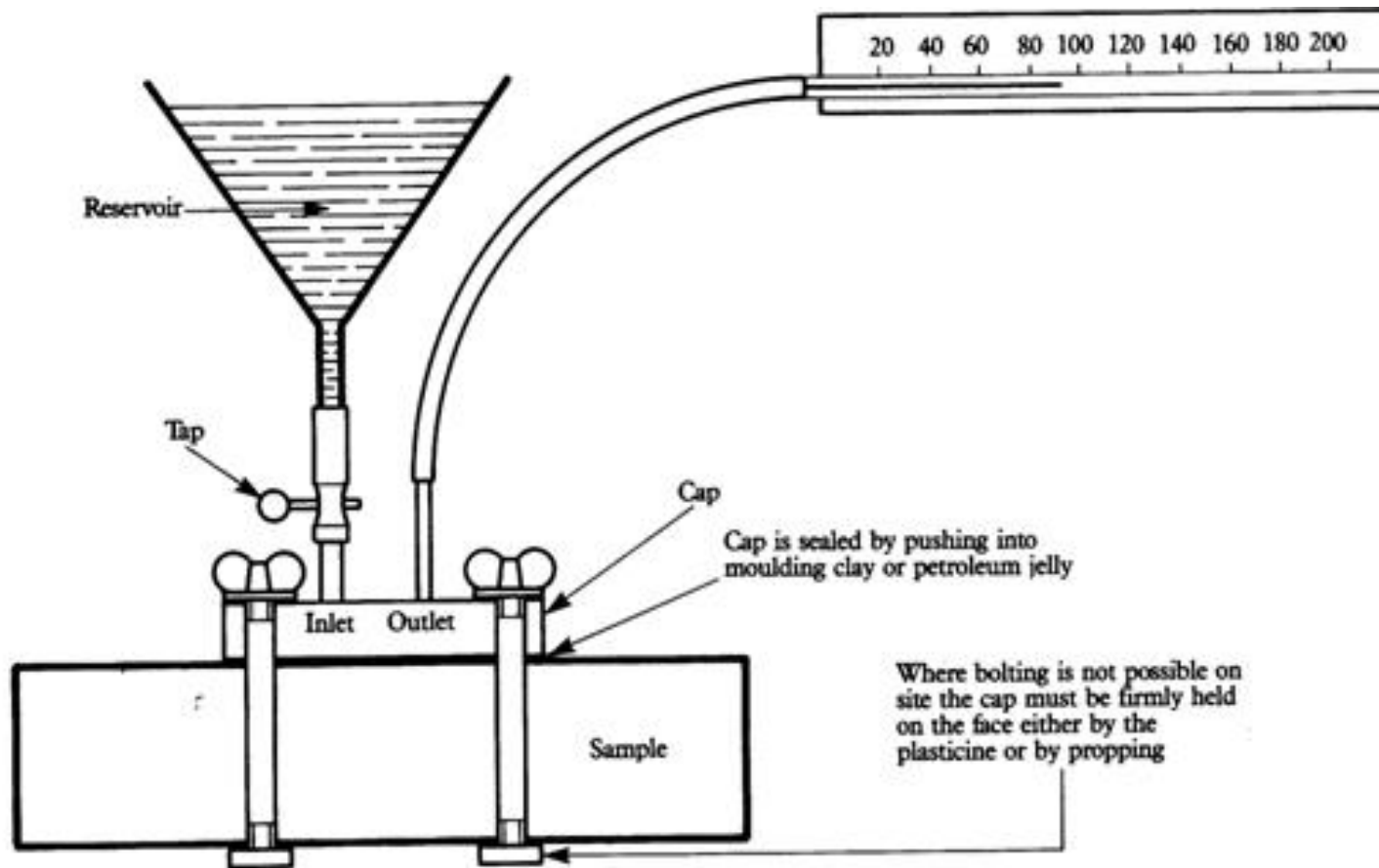


# Initial Surface Absorption Test (ISAT)

- Involves measurement of **rate of flow** of water per unit area into a concrete surface when subjected to a constant head of 200mm.
- A watertight cap is sealed to the concrete surface to provide water contact of at least 5000mm<sup>2</sup> & connected to a reservoir & calibrated horizontal capillary tube & scale.
- At specified intervals from start of test, the tap between cap & reservoir is closed & movement of water in capillary is measured.
- The test method is covered by BS 1881-208.



# Initial Surface Absorption Test (ISAT)



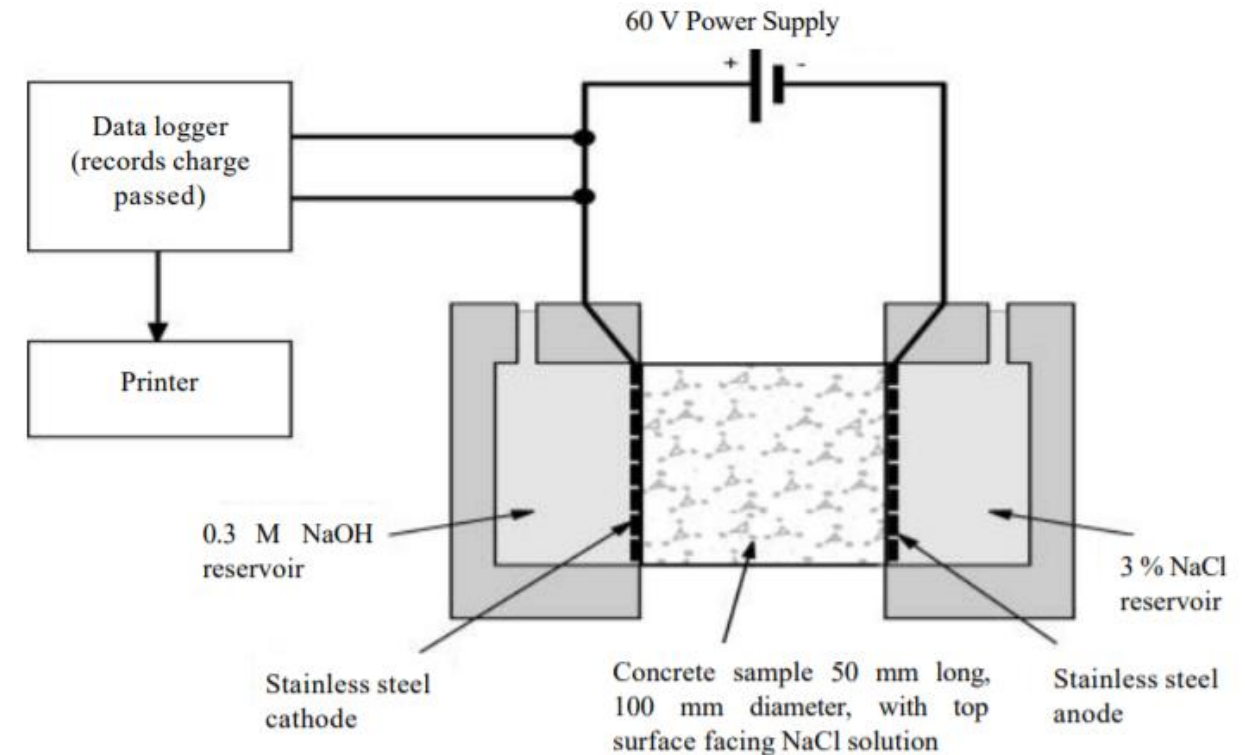
# Procedure

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# Rapid Chloride Penetration Test (RCPT)

- Indication as concrete ability to resist chloride ion penetration.
- Accordance with **ASTM C1202** (2012) to determine the chloride ions charge.
- Carried out on saturated surfaced diameter of **100 × 50 mm** cylindrical.
- The cell that was connected to the **positive terminal** of the power supply was filled with 0.3 N **sodium hydroxide** (NaOH) solution.
- The cell that was connected to the **negative terminal** was filled with 3% **sodium chloride** (NaCl) solution.
- subjected to a 60 V applied DC voltage for **6 hours** and the current reading was recorded every 30 minutes using data logger.

# Rapid Chloride Penetration Test (RCPT)



# RCPT Ratings

Charge Passed (coulombs)	Chloride Ion Penetrability	Typical concrete type (Joshi and Chan, 2002)
>4,000	High	High w/c ratio (>0.6) conventional OPC concrete
2,000 – 4,000	Moderate	Moderate w/c ratio (0.4-0.5) conventional OPC concrete
1,000 – 2,000	Low	Low w/c ratio (<0.4) conventional OPC concrete
100 – 1,000	Very Low	Latex-modified concrete, internally sealed concrete
<100	Negligible	Polymer-impregnated concrete, polymer concrete

**ASTM C1202**

# Rapid Chloride Penetration Test (RCPT)

## Advantages

- Is relatively quick—can be used for quality control
- Has simple and convenient setup and procedures
- Provides results that are easy to interpret
- Correlates well with 90-day chloride ponding test

## Disadvantages

- May not represent the true permeability (or potential permeability) for concrete that contains supplementary cementitious materials or chemical admixtures
- May allow measurements before a steady state is achieved
- Can cause physical and chemical changes in the specimen, resulting in unrealistic values
- May not be suitable for concretes that contain conducting materials (such as steel or carbon fibers)
- Has low inherent repeatability and reproducibility



# Cathodic Protection

- Cathodic protection is **fundamental** to preserving a **pipeline's integrity**.
- Cathodic protection is a method of **corrosion control** that is achieved by supplying an external direct current that neutralizes the natural corrosion current arising on the pipeline at coating defects.
- A simple method of protection connects the metal to be protected to a more easily corroded "**sacrificial metal**" to act as the anode.



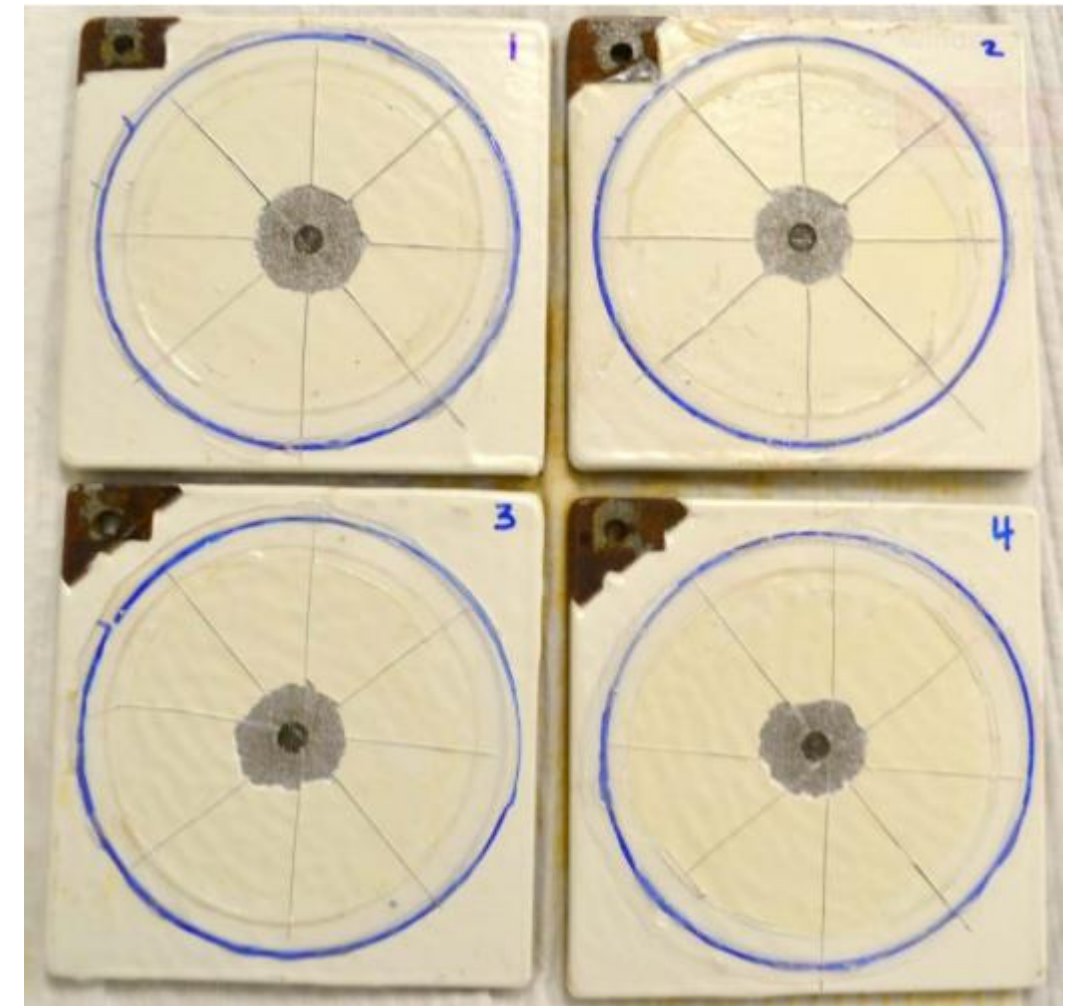


# Cathodic Protection

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- Current required to protect a pipeline is dependent on the environment and the number and size of the coating defects.
- Clearly, in a particular environment, the greater the number and size of coating defects, the greater the amount of current required for protection.
- Coating plays an integral part in the functioning of a pipeline's cathodic protection system. Where a coating system has badly deteriorated, cathodic protection requirements and costs can increase exponentially.

# Cathodic Disbondment Testing (CDT)



# Procedure

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# Testing for Reinforcement Corrosion

- Reinforcement corrosion may lead to structural weakening due to :
  - loss of steel cross-section (bars or strands);
  - cracking & spalling;
  - internal delamination ( concrete fracture plane occurs at the level of the corroding steel mesh).





# Testing for Reinforcement Corrosion

- To identify the **risk** of corrosion (not apparent at the surface);
- To identify the **extent** of corrosion (partially observable at the surface);
- To assess the **integrity** of a structure which may have experienced corrosion.



# Theory of Corrosion

- Steel reinforcement bars are **protected** by alkaline environment (passivity) provided by the pore fluid in hardened concrete.
- Corrosion starts when passivation is broken down by **carbonation** or ingress of **chlorides**.
- Corrosion is either **localized** or **generalized** form (generalized form is most disruptive due to expansion of rusting steel).



# Half-Cell Potential

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- Half Cell corrosion mapping is an **effective method** for assessing the severity of corrosion activity in concrete structures.
- Most well-known procedure to identify the likelihood of **active corrosion**
- The test does **not provide** any information about the kinetics of **corrosion activity**.
- **Easy and cost-effective** procedure of the test makes it popular among engineers and structure inspectors.



# Half-Cell Potential

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- Involves measuring the potential of an **embedded reinforcement bar** relative to a reference half-cell placed on the concrete surface.
- The difference in potential between concrete surface and steel is a good indicator of **current flow**.
- The **electrochemical process** produces an **electric current**, which is measurable as an electric field on the surface of the concrete.
- Copper/Copper Sulphate or Silver/Silver Chloride cells are typically used (See TRRL Application Guide 9 (1991) for guidance)
- Several standard associations have standardized the test procedure including the ASTM C 876, UNI 10174 and RILEM TC 154.

# Half-Cell Potential

Potential Value	Possible Corrosion Rate
$\leq -0.20 \text{ V}$	90% probability of no corrosion
$-0.20 \text{ to } -0.35 \text{ V}$	Corrosion activity uncertain
$> -0.35 \text{ V}$	more than 90% probability of corrosion

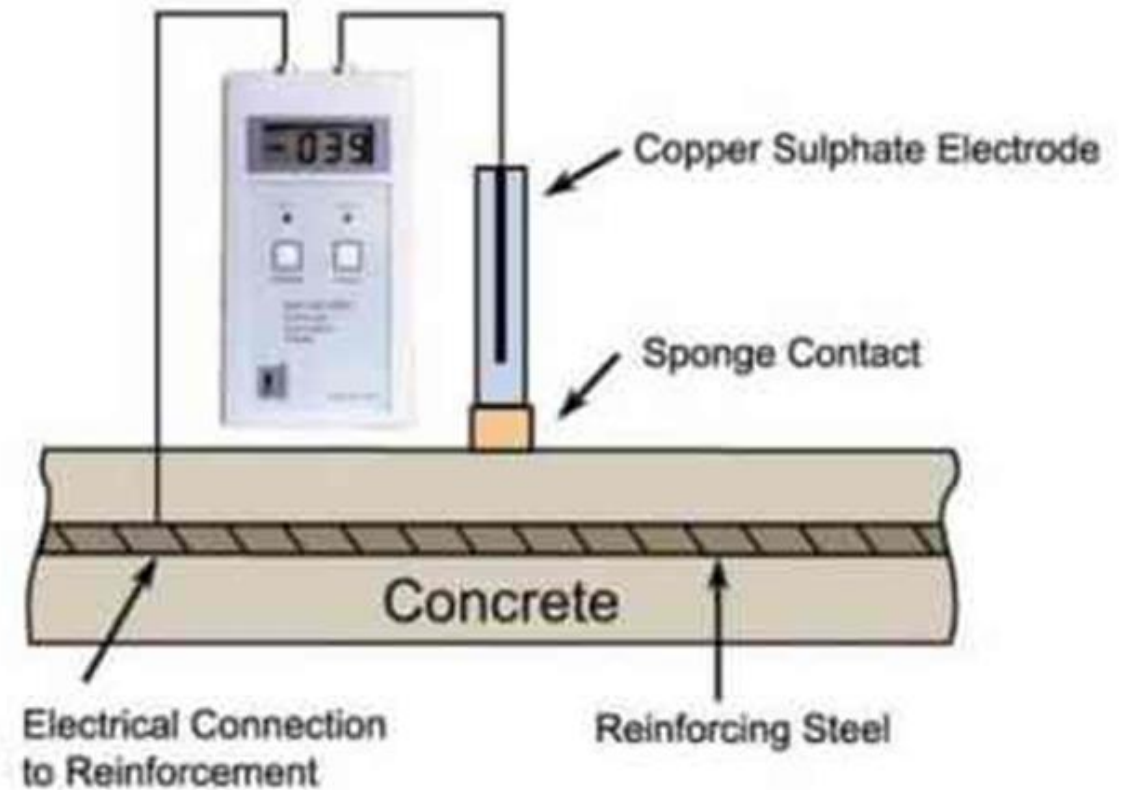
**Value of Half-cell Potential Test Versus the possible Rate of Corrosion of Steel bars in Concrete**

# Half-Cell Potential

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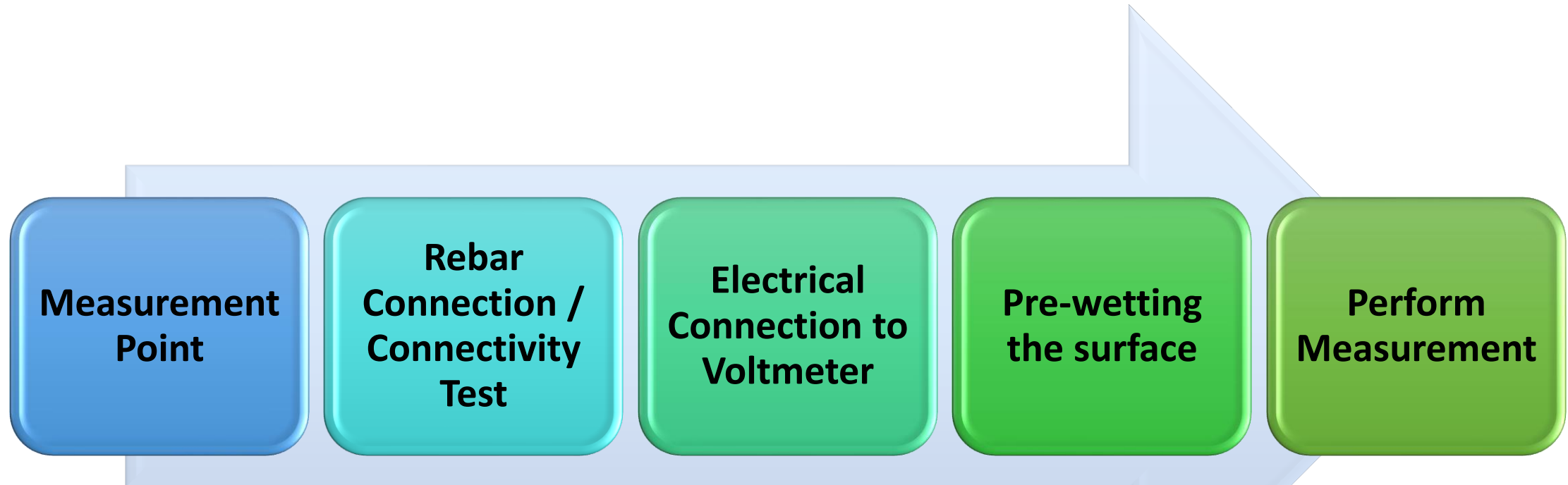
- Simple equipment enables **NDT survey** to produce **iso-potential contour maps** of the surface of concrete member.
- Zones of varying degrees of corrosion risk may be identified from these 'maps'.
- Half-cell method **cannot indicate actual** corrosion rate.
- Drilled hole may be needed to make electrical contact with the reinforcement bar.
- The positive terminal of the voltmeter is connected to exposed rebars and the negative terminal (common) to reference half-cell.

# Half-Cell Potential



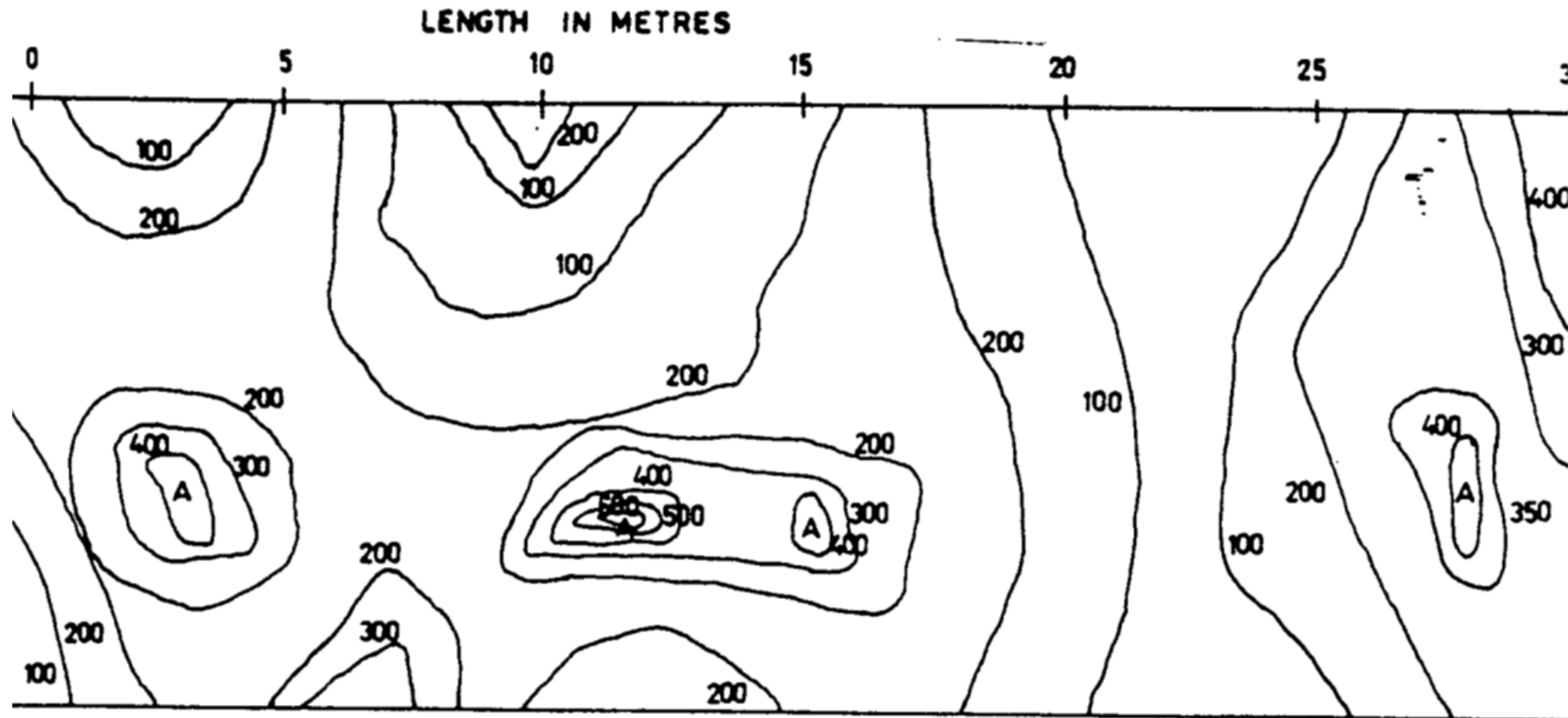
# Procedure

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ASTM C876

# Equipotential Contours



# Influencing Parameters

- Electrical resistivity of concrete
- Density of concrete
  - (dense concrete versus porous concrete)
- Cover thickness
- Epoxy coatings
- Oxygen Availability
- Environmental conditions
  - (such as moisture and humidity)

“The higher the electrical resistivity is, the less accurate the half-cell potential measurements from the concrete surface will be.”

“A decrease in oxygen concentration at the surface of the steel reinforcement will result in a more negative corrosion potential reading.”



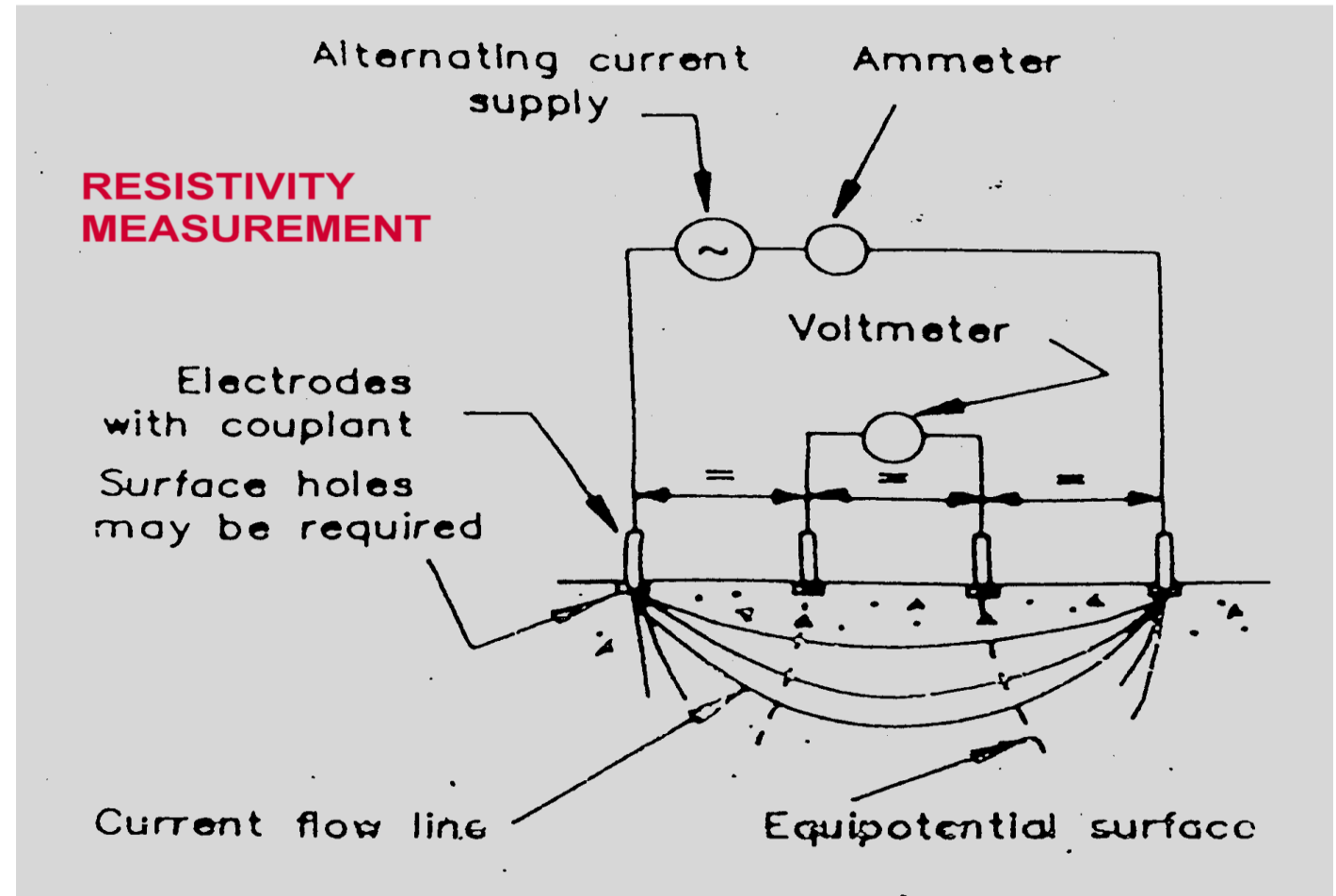
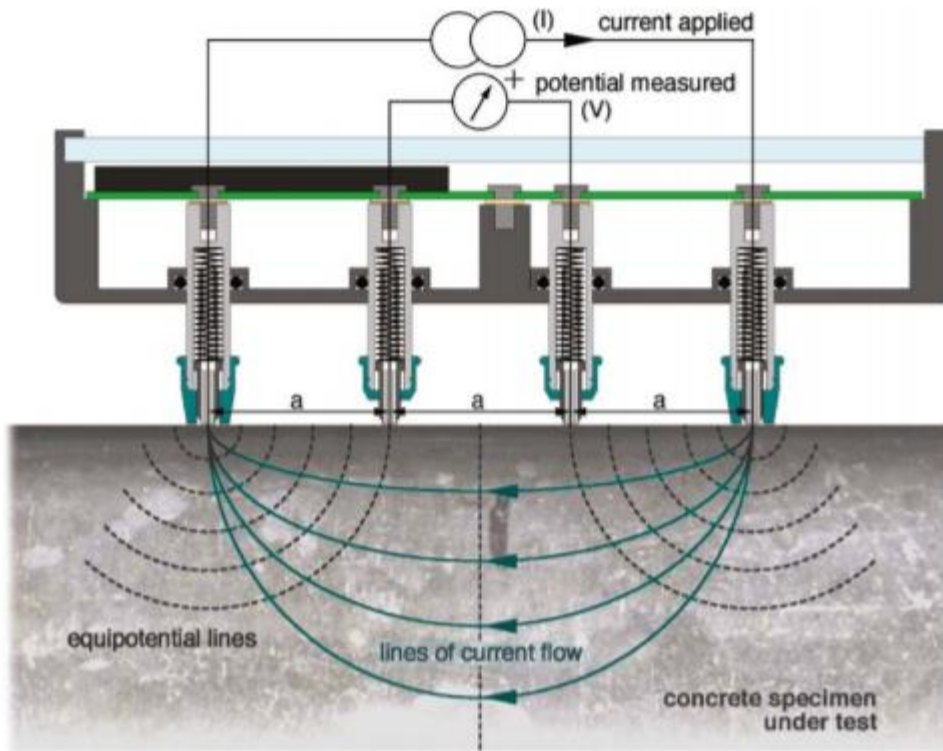
# Resistivity Meter

- The corrosion of steel in concrete is an **electrochemical** process that generates a **flow of current**.
- The resistivity of the concrete influences the flow of this current.
- The **lower** the electric resistance, the more easily corrosion current flows through the concrete and the greater is the probability of corrosion.
- The resistivity of concrete is a good indication of the probability of corrosion.
- Resistivity meter can be used to **assess the probability of corrosion** of steel reinforcement embedded in concrete.
- It is a very **simple technique** and can be **adopted easily** in the field.

# Resistivity Meter

- Resistivity meter is very **handy and portable** equipment. The concrete resistivity is displayed on an LCD.
- Available with non-volatile memory and colored graphic display from which data can be transferred on PC.
- To measure the resistivity, metallic probes are placed over the concrete surface. A known current is passed on the outer probes and resulting potential drop between inner probes is measured.
- The resistance is computed by dividing the potential drop by the current.
- A conductive gel is used between probe and concrete surface to make effective contact.

# Resistivity Measurement



# Resistivity Meter



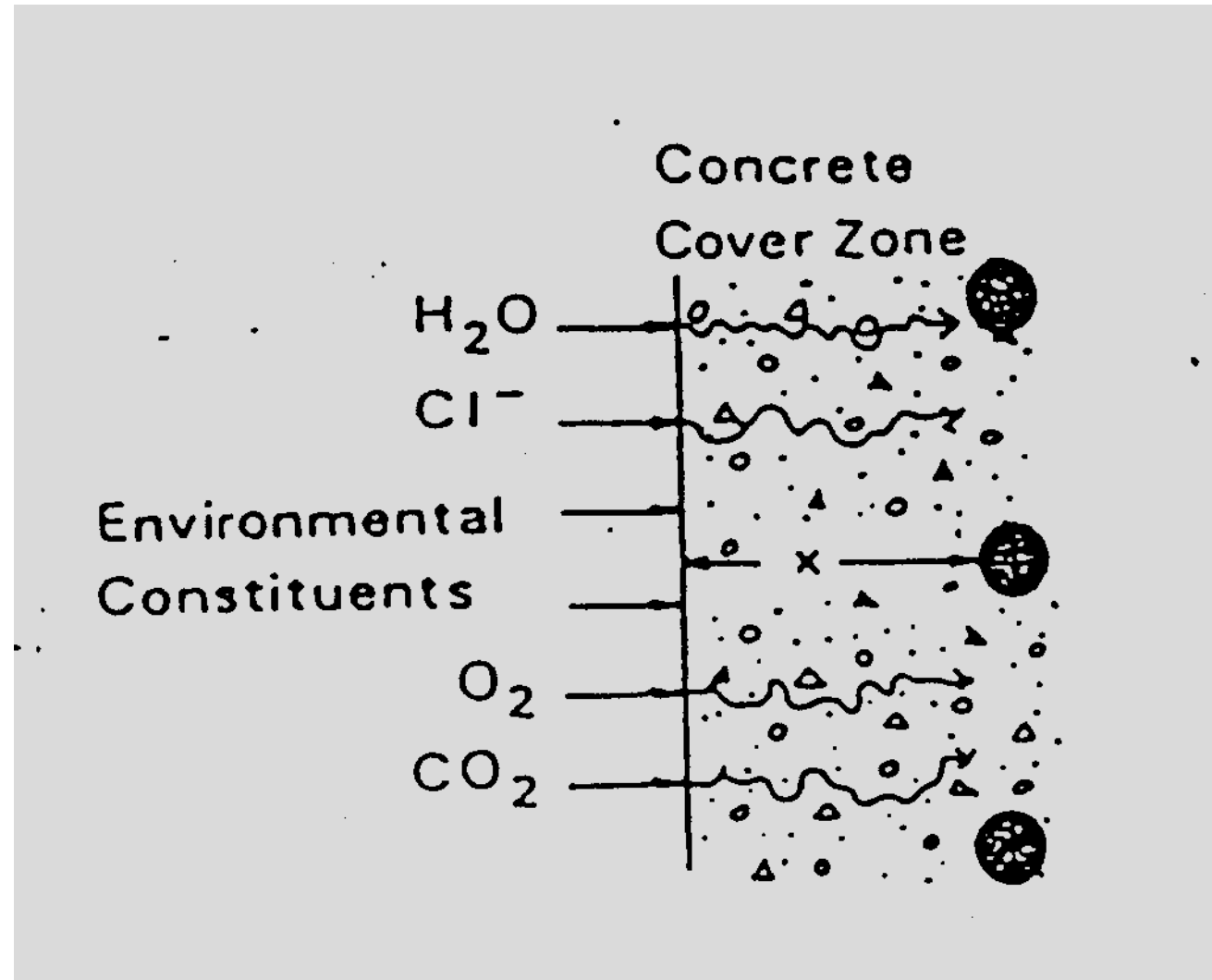
# Resistivity Meter

Concrete Resistivity (ohm/cm)	Rate of Corrosion
>20,000	Negligible
10,000 –20,000	Low
5,000 –10,000	High
< 5,000	Very High

**Resistivity Level Versus Possible Corrosion Rate of Steel Reinforcements in Concrete**



# Carbonation & Chloride Attack



# Carbonation

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- Carbonation is the effect of  $\text{CO}_2$  in the air on Portland Cement products, mainly  $\text{Ca}(\text{OH})_2$ , in the presence of moisture.
- If concrete is of good quality (**dense & impermeable**), further ingress of  $\text{CO}_2$  is prevented.
- In poor quality concrete, carbonation will not be halted at the surface but will proceed through the depth, resulting in **lowering of pH** from 11.5 to 9 or less.

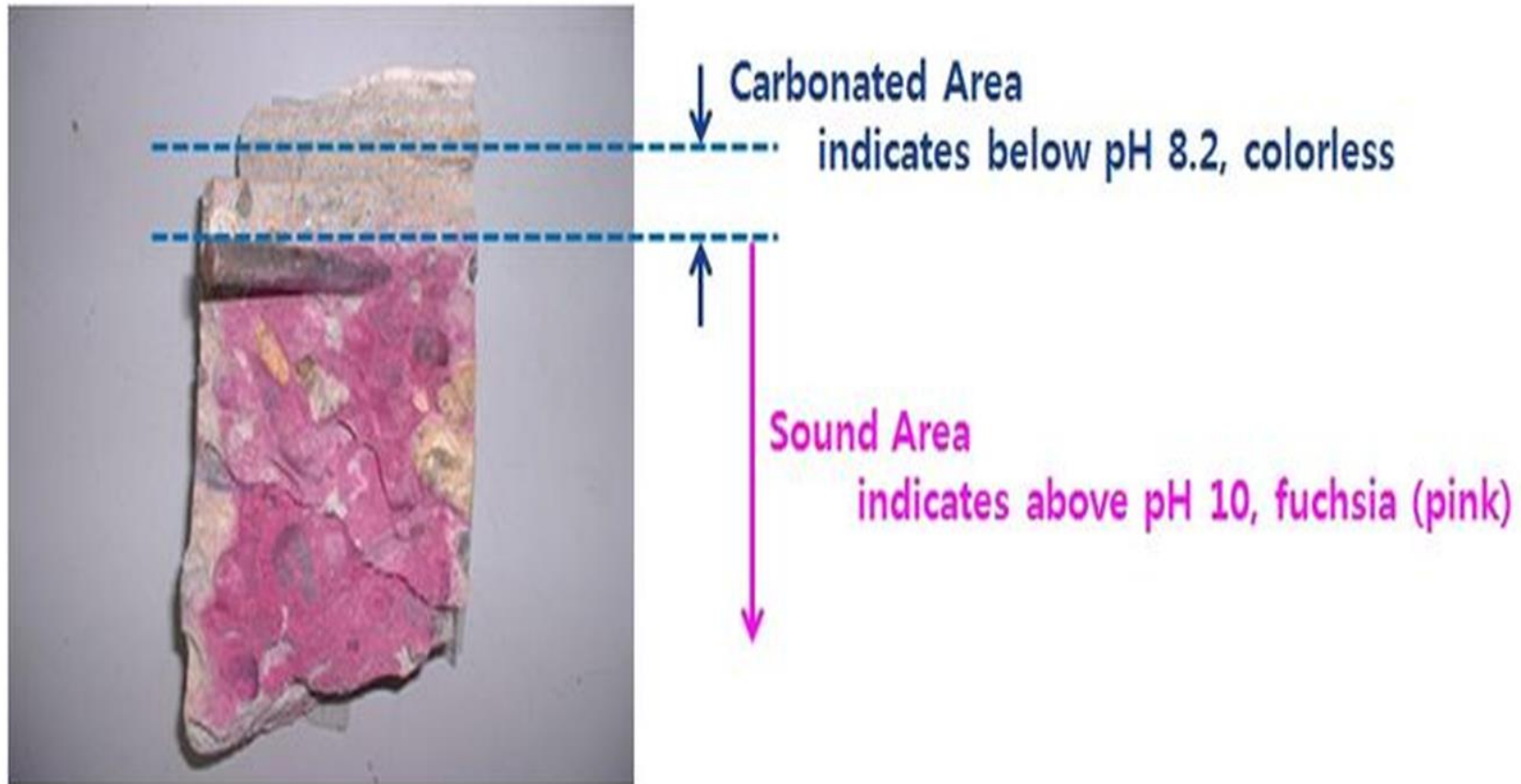


# Carbonation Depth

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- The loss of alkalinity associated with carbonation of surface concrete may be detected by the use of **phenolphthalein** sprayed onto a freshly exposed or broken surface.
- A **quick & simple** approach which gives a visual indication of the position of the **depassivation** front relative to the steel.
- Test may also be conducted on core samples.
- Solution turns pink where  $\text{ph} > 9$ .

# Carbonation Test



# Chloride Attack

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- Apart from being used in admixtures, chloride ions may be introduced into concrete by other means :
  - Diffusion into mature concrete;
  - Accidental inclusion as contaminants in aggregates;
  - Penetration by de-icing slats, industrial brines, marine spray, fog etc.

# Chloride Ion Penetration

- May be identified by **chemical analysis** of powdered drillings obtained from various depths below the concrete surface.
- Simple site tests (Hach & Quantab) are useful in providing preliminary results to establish the need for more sophisticated laboratory methods.
- Results expressed in terms of **% chloride ion** by weight of cement (0.35% is critical). Initial results by weight of concrete may be OK except for borderline cases.
- Detailed chemical analysis provided by BS1881 Part 124 (1988) and Concrete Society TR 32(1989).

# Portable Chloride Test Kit





# Chloride Profile Grinder





# Chloride Categorization

- Categorization applies to chlorides which were present in the original concrete mix.
  - a) 'Low' –with chloride ion content of up to 0.4% by weight of cement
  - b) 'Medium' –from 0.4% to 1%
  - c) 'High' –above 1%
- Lower limits may apply if chloride has penetrated from outside the hardened concrete

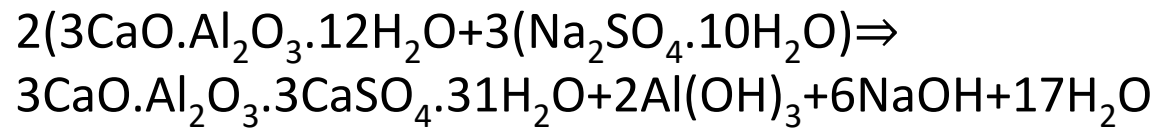
**BRE Digest 264 Guide for Chloride Categorization**

# Chloride Categorization

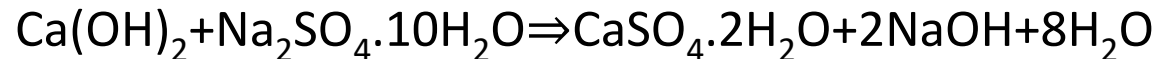
- When cement content values are not known, the chloride content is expressed in terms of weight of chloride by weight of concrete.
  - a) Up to 0.02% : level insignificant
  - b) Up to 0.05% : chlorides unlikely to have been added to the mix, low risk category.
  - c) 0.05% - 0.15% : medium risk category.
  - d) Above 0.15% : high risk category.

# Sulphate Attack

The formation of **ettringite** (calcium aluminate trisulfate), resulting in an increase in solid volume, leading to **expansion** and **cracking**:



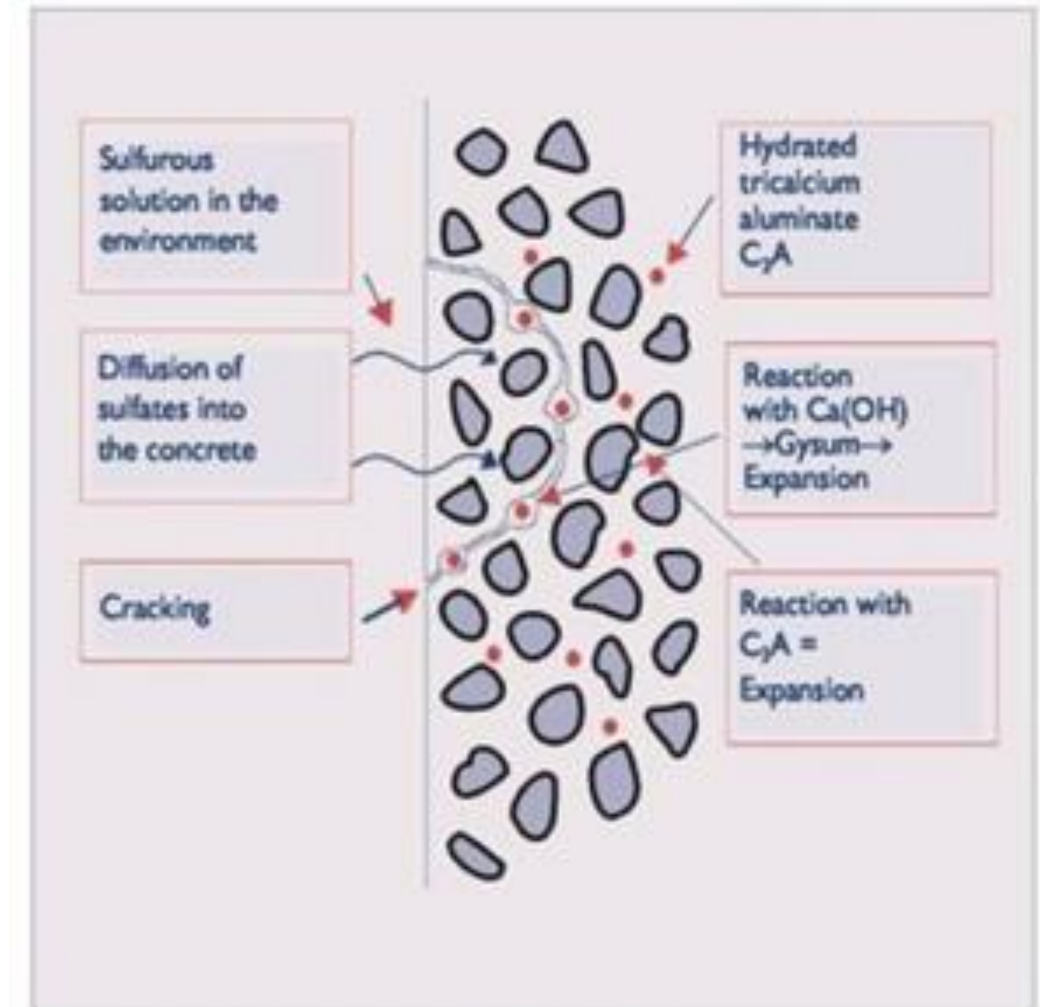
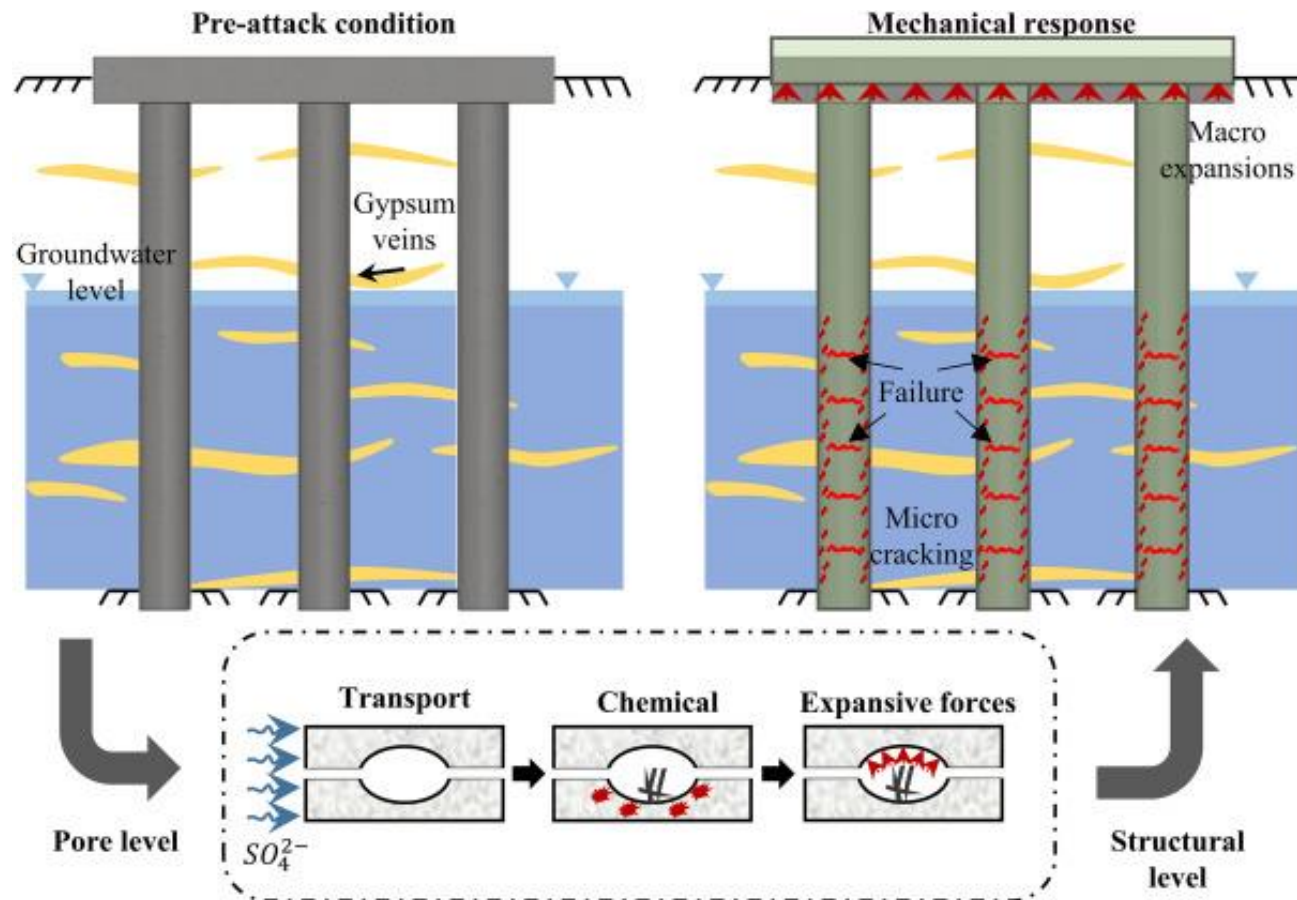
The formation of gypsum (calcium sulfate dihydrate), leading to softening and loss of concrete strength:



**\*Ettringite and gypsum formation lead to deterioration of concrete due to sulphate attack.**

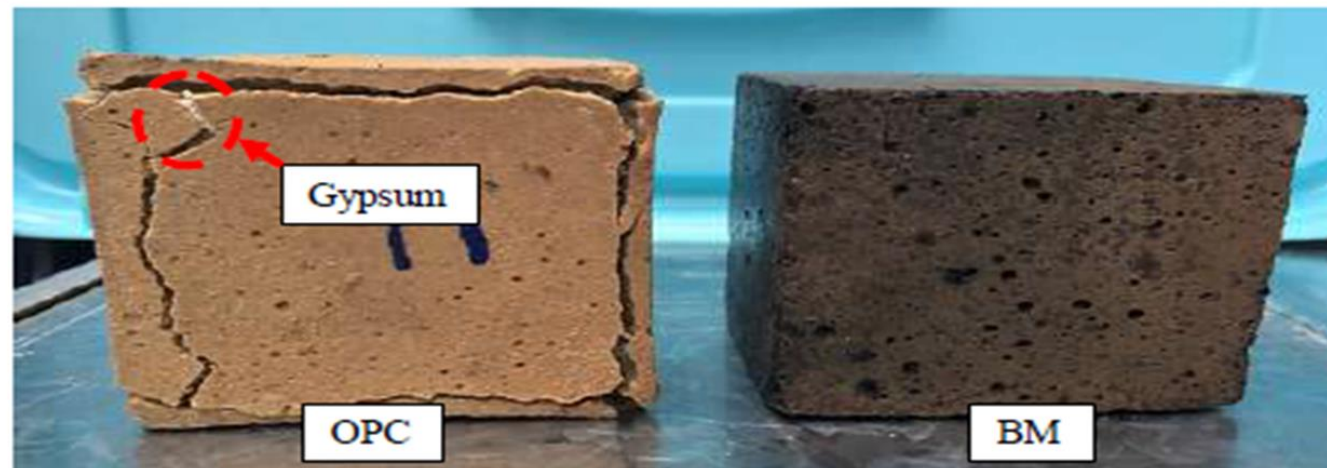
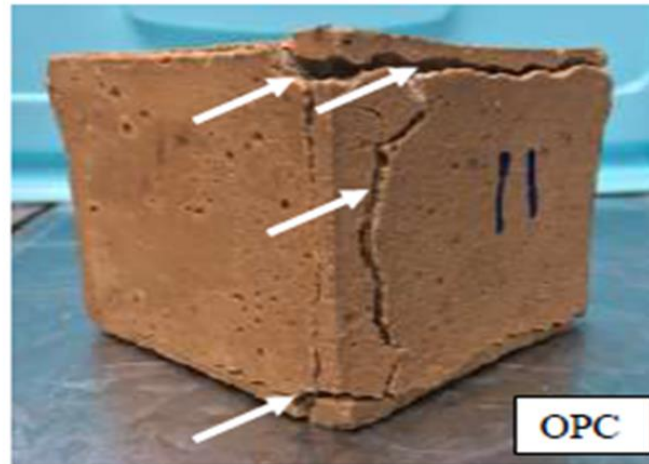


# Mechanism





# Sulphate Attack



# Concentration limit

Exposure	Concentration of Water-Soluble Sulphates in Soil (%)	Concentration of Water-Soluble Sulphates in Water (ppm)
Mild	< 0.1	< 150
Moderate	0.1 to 0.2	150 to 1500
Severe	0.2 to 2	1500 to 10000
Very Severe	> 2	> 10000



# Thank You



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